Polyatomic Ion List

H₃O+	hydronium	NH_4 +	ammonium
Hg ₂ ²⁺	mercurous	OH-	hydroxide
NO_3	nitrate	NO_2^-	nitrite
O_2^{2-}	peroxide	CO ₃ ²⁻	carbonate
HCO ₃ -	hydrogen carbonate	SO ₄ ²⁻	sulfate
HSO ₄ -	hydrogen sulfate	SO ₃ ²⁻	sulfite
HSO ₃ -	hydrogen sulfite	MnO_4	permanganate
CrO ₄ ²⁻	chromate	Cr ₂ O ₇ ²⁻	dichromate
PO ₄ ³⁻	phosphate	HPO ₄ ²⁻	hydrogen phosphate
$H_2PO_4^-$	dihydrogen phosphate	PO ₃ ³⁻	phosphite
HPO ₃ ²⁻	hydrogen phosphite	$H_2PO_3^-$	dihydrogen phosphite
$C_2H_3O_2^{-1}$	acetate	NH_4^+	ammonium
$C_2O_4^{2-}$	oxylate	CN-	cyanide
CIO ₄ -	perchlorate	CIO ₃ -	chlorate
CIO ₂ -	chlorite	CIO-	hypochlorite
BrO ₃ -	bromate	BrO ₂ -	bromite
CNO-	cyanate	SCN-	thiocyanate

Learn more about polyatomic ions:

Source: www.physicsclassroom.com

Chemistry Tutorial → Chapter 3 → Lesson 3c

